SHORT COMMUNICATIONS

Stain Film Formed Photochemically on Silicon

By Yasuo Nakai

(Received May 21, 1965)

It is known that stain film can contribute to the stabilization of transistors.1) The stain film has been obtained hitherto by reaction with hydrogen fluoride-nitric acid²⁾ or by anodic oxidation in hydrogen fluoride solution.3)

In this experiment, it was found that the formation of stain film can be effected by the photochemical reaction of HF-O₃ with silicon crystal. For example, oxygen gas (1200 cc./ min.) containing 5% ozone (in weight) and HF-N₂ gas (500 cc./min.) was brought into contact with the (111) surface of 2- to 80chm-cm. n-type silicon crystals for 0.5 to 2 hr. at room temperature and under irradiation from a high-pressure mercury lamp (1 kW.) through filters transmitting at 5800, 5460, 4358, 4047 and 3650Å. Ozone was obtained by a silent discharge. The mixture of HF-N2 gas was obtained by bubbling nitrogen gas through a 24 N hydrogen fluoride solution at 25°C.

It was found to be necessary, for the formation of stain film, to irradiate the silicon surface with photons of energy higher than approximately 2.7 eV.

Infrared, optical absorption, electron spin resonance and solid mass spectra were investigated in studying the physicochemical properties of stain film.

¹⁾ K. Oho and K. Koso, ECL. Tech. Jour., 10, 1741 (1951): S. Y. Ellis, J. Appl. Phys., 28, 1262 (1957).
2) D. R. Turner, "The Electrochemistry of Semiconductors." Ed. by P. T. Holms, Academic Press, New York (1962).

³⁾ R. J. Archer, J. Phys. Chem. Solids, 14, 104 (1960); A. Uhlier, Bell. Syst. Tech. J., 35, 333 (1956).

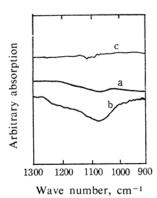


Fig. 1. The infrared spectra of stain film (a) before and (b) after thermal treatment at 250°C for 10 hr., and (c) of silicon crystal.

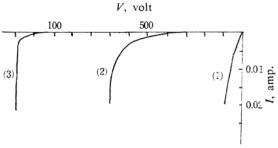


Fig. 2. Reverse characteristic (I-V) curve of a p^+-n diode on which stain film was photochemically formed.

(1) Before heat treatment (2) after being kept at 250°C for 2 hr. in air, and (3) after being kept at 250°C for 10 hr. in air.

Figure 1 shows the infrared spectrum of stain film. The band at $1080 \,\mathrm{cm^{-1}}$, whose intensity is weak before heat treatment (as is shown by curve 1a), appears strongly after being kept at $250^{\circ}\mathrm{C}$ for approximately 10 hr. in air (curve 1b). The band at $1080 \,\mathrm{cm^{-1}}$ is assigned to the Si-O stretching mode; its shape is similar to that of amorphous silica.

The electron-spin resonance-absorption curve of stain film at an operation microwave frequency in the vicinity of 9.2 kMc./sec. exhibits three peaks ($g_1 = 2.0068$, $g_2 = 2.0048$ and $g_3 = 2.0000$) at 77°K. Two optical absorption bands were observed, at 7500Å and 5000Å. Silicon, oxygen and fluorine atoms were detected in stain film by solid mass spectroscopy, but its molecular formula is not clear at present.

Stain film was photochemically formed upon the junction of mesa p^+-n diodes. A typical variation of the reverse *I-V* curve with aging is shown in Fig. 2. Diodes had a very inferior reverse characteristic just after forming stain film, but the breakdown voltage became stable and the drift decreased after they had been kept at 250°C for 10 hr. in air.⁴⁾

Central Research Laboratory Tokyo Shibaura Electric Co., Ltd. Kawasaki-shi, Kanagawa

⁴⁾ Y. Nakai, T. Owada, S. Fukasaku, M. Urakabe and T. Oi, J. Electrochem. Soc., to be published.